

A Convenient and Improved Preparation of Cyano Phosphates

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Abstract: Cyano phosphates derived from aldehydes or ketones are prepared in good yields using lithium cyanide in DMF and diethyl chlorophosphate as phosphorylating agent.

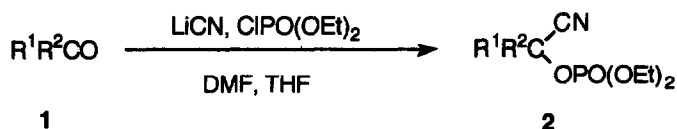
Introduction

Cyanohydrin diethyl phosphates are versatile intermediates in synthetic organic chemistry¹ specially: (a) in the preparation of saturated² and α,β -unsaturated³ nitriles from carbonyl compounds by reductive elimination of the phosphate group, (b) for the regiospecific Friedel-Crafts type arylation of aromatic compounds⁴, (c) in the synthesis of γ -hydroxy α,β -unsaturated nitriles^{4b,5} and β -cyano α,β -unsaturated ketones⁵ by [3,3]-sigmatropic rearrangement of α,β -unsaturated cyano phosphates, (d) as acyl anion equivalents⁶ in the case of aldehydes derivatives, and (e) in the synthesis of indole derivatives⁷. We were interested in the synthesis of saturated nitriles from saturated ketones by reduction of the corresponding cyano phosphates with samarium iodide^{2b}. The general method described for the preparation of diethylphosphoric esters of cyanohydrins is the treatment of carbonyl compounds either with a three-fold excess of lithium cyanide and *O,O'*-diethyl phosphorocyanidate [(EtO)₂P(O)CN, DEPC]¹⁻⁷, or with DEPC in the presence of a catalytic amount of lithium diisopropylamide⁸. We have found a more simple and cheaper way to prepare cyano phosphates using as reagents lithium cyanide and diethyl chlorophosphate.

Results and Discussion

The reaction of carbonyl compounds **1** with a slight excess of a solution of lithium cyanide (1.5 eq) in *N,N*-dimethylformamide⁹ and diethyl chlorophosphate (1.2 eq) in THF at room temperature gave cyanohydrin diethyl phosphates **2b-c**, **2g**, **2h** and **2j-m** in good yields (Scheme 1 and Table 1).

The process was monitored by GLC and took place in less than one hour and with similar yields than in the DEPC method¹⁻⁷, and compounds **2** were obtained in most cases as crude products in >93% of purity (GLC and 300MHz ¹H-NMR). In the case of α,β -unsaturated carbonyl compounds such as (*E*)-crotonaldehyde (**1b**) and methyl vinyl ketone (**1g**) only the corresponding 1,2-addition products were formed. However,



Scheme 1

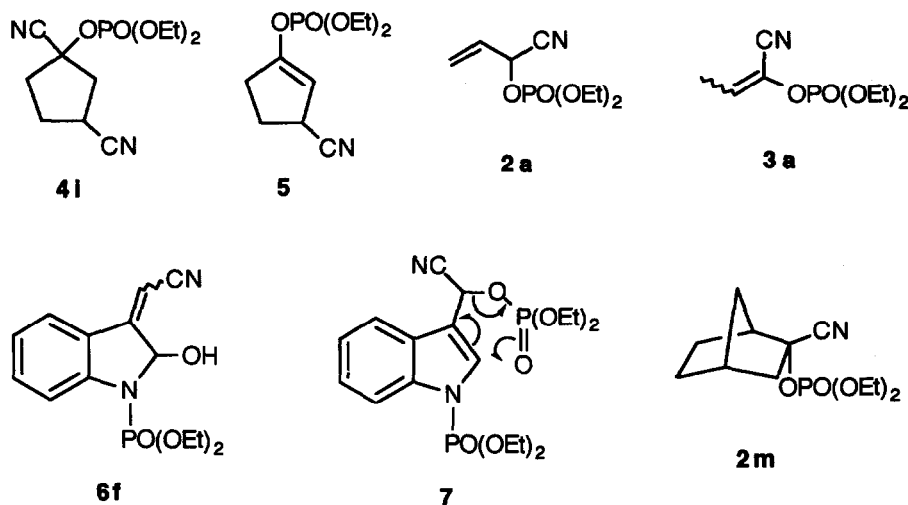
Table 1. Synthesis of Cyano Phosphates 2-4 and 6.

| Carbonyl compound | | | Reaction time ^a | Product | | |
|-------------------|---|----------------|-------------------------------|-----------------|------------------------|---------------------|
| no. | R ¹ | R ² | | no. | yield (%) ^b | bp(°C) ^c |
| 1a | CH ₂ =CH | H | 10 min | 3a ^d | 66 | 90 |
| 1b | (<i>E</i>)-MeCH=CH | H | 10 min | 2b | 85 | 105 ^e |
| 1c | <i>t</i> -Bu | H | 15 min | 2c | 85 | 100 |
| 1d | Ph | H | 30 min | 2d | 88 | 135 ^f |
| 1e | PhCH ₂ | H | 1 h | 2e | 93 | 160 |
| 1f | 3-indolyl | H | 2 h ^{g,h} | 6f ⁱ | 72 | -j,k |
| 1g | CH ₂ =CH | Me | 10 min ^g | 2g | 81 | 90 ^l |
| 1h | Et | Et | 10 min ^g | 2h | 67 | 110 |
| 1i | -(CH ₂ CH ₂ CH=CH)- | | 15 min | 4i ^m | 40 | 160 ⁿ |
| 1j | -(CH ₂) ₅ - | | 40 min | 2j | 90 | 110 ^o |
| 1k | -p | | 45 min | 2k | 90 | 170 |
| 1l | Ph | Me | 1 h | 2l | 84 | 140 ^q |
| 1m | -r | | 10 min | 2m | 51 ^s | 120 |

^aThe reaction was carried out at room temperature. ^b Isolated crude yield based on compound 1. All crude compounds 2-4 and 6 were >93% pure (GLC and 300MHz ¹H-NMR). ^c Kugelrohr (4 Torr). ^d *Z/E* ratio: 9/1 (GLC). ^e Lit. 4b. ^f Lit. 6a: 149°C (3 Torr). ^g At 0°C. ^h LiCN was slowly added during 30 min. ⁱ *Z/E* ratio: 65/35 (¹H-NMR). ^j Lit. 7. ^k *R_f* 0.38 for the *Z*-isomer and 0.28 for the *E*-isomer, hexane/ethyl acetate: 1/1. ^l Lit. 4b: 77-83°C (2 Torr). ^m *Cis/trans* ratio: 1/1 (¹H-NMR). ⁿ *T_r* 10.30 and 10.39 min (see Experimental). ^o Lit. 8. ^p 1,4-Cyclohexanedione monoethylene ketal. ^q Lit. 2a. ^r Norbornanone. ^s After column chromatography on silica gel (hexane/ethyl acetate).

2-cyclopentenone (**11**) gave the 1,4-addition product **4i** as *cis/trans* mixture of diastereoisomers in *ca.* 1/1 molar ratio (Table 1), while the DEPC method gave the vinyl phosphate **5** under the similar reaction conditions⁵. When the reaction was carried out with acrolein (**1a**) the unsaturated cyano phosphate **3a** was exclusively obtained as a *Z/E* mixture of diastereoisomers in 9/1 molar ratio (Table 1), this product arises from the isomerization of compound **2a**. On the other hand, the reaction of acrolein (**1a**) with lithium cyanide and DEPC under the same reaction conditions gave a mixture 3/1 of compounds **2a** and **3a**.

The treatment of 3-indolecarbaldehyde (**1f**) as above at 0°C afforded a mixture of (*Z*)- and (*E*)-3-cyanomethyleneindolines **6f** in 72% yield, in 65 and 35% ratio respectively. The same result was obtained with the DEPC method, product (*E*)-**6f** being the major isomer⁷. Compounds **6f** were formed by [3,3]-sigmatropic rearrangement of the intermediate cyanophosphate **7**. In the case of norbornanone (**1m**) only one diastereoisomer (**2m**), presumably with the cyano group in the *exo* position, was obtained.



The procedure of cyanophosphorylation described here is a good method for the synthesis of synthetic useful cyano phosphates and allows the use of diethyl chlorophosphate instead of diethyl cyanophosphate in stoichiometric amounts.

Experimental Part

General. - IR spectra were obtained as films in a Pye Unicam SP3-200 spectrophotometer as neat liquids. ¹H, ¹³C, and ³¹P NMR spectra were recorded on a Bruker AC-300 spectrometer with SiMe₄ as internal standard or H₃PO₄ 85% and using CDCl₃ as solvent. ¹³C-NMR assignments were made on the basis of DEPT experiments. MS spectra were measured in a Hewlett-Packard 5988A by GLC injection (EI, 70eV). Elemental analyses were performed by the Microanalyses Service of the University of Alicante. Chromatographic analysis (GLC) were determined with a Hewlett-Packard HP-5890 instrument equipped with a 25 m WCOT capillary column (0.22 mm diam., 0.2 μm film thickness OV-101 stationary phase) using nitrogen (2 ml/min) as the carrier gas, T_{injector}=270°C, T_{column}=100°C, and 100-270 (10°C/min). Thin layer chromatography (TLC) was carried

out on Schleicher & Schuell F1500/LS 254 plates coated with a 0.2 mm layer of silica gel, using a mixture of hexane/ethyl acetate as eluant, and revealed, in general, with iodine. Column chromatography was performed using silica gel 60 of 70-270 mesh and hexane/ethyl acetate as eluant. All starting materials were commercially available (Aldrich, Fluka) of the best grade and were used without further purification. THF was dried with LiAlH_4 under argon atmosphere.

Preparation of Compounds 2-6. General Procedure.— To a solution of carbonyl compound **1** (0.5 mmol) and diethyl chlorophosphate (0.09 ml, 0.6 mmol) in dry THF (4 ml) was added a 0.5 M solution of lithium cyanide⁹ (1.5 ml, 0.75 mmol) in DMF and the mixture was stirred at room temperature (see Table 1). The reaction mixture was quenched with water and extracted with a mixture of hexane and ethyl acetate (1/1). The organic layer was washed three times with brine, dried over anhydrous sodium sulfate and evaporated (15 Torr) to yield crude pure cyano phosphates **2-4** and **6**, which were purified by distillation or column chromatography. Yields and bp values are given in Table 1, R_f values (hexane/ethyl acetate: 2/1), spectral, and analytical data for new compounds follow.

(Z/E)-2-(Diethylphosphonoxy)-2-butenenitrile (3a): R_f 0.48; ν_{\max} 3060, 1630 (CH=C), 2225 (CN), 1280 (P=O), and 1020 cm^{-1} (P-O); δ_{H} 1.41 (t, $J=7.0\text{Hz}$, 6H, $2\times\text{CH}_3\text{CH}_2$), 1.89 (Z), 1.96 (E) (2dd, $J=7.5, 2.5\text{Hz}$, 3H, CH_3CH), 4.26 (m, 4H, $2\times\text{CH}_2$), 6.11 (qd, $J=7.0, 1.0\text{Hz}$, 0.9H, CH), 6.38 (qd, $J=7.5, 2.5\text{Hz}$, 0.1H, CH); δ_{C} 11.64 (Z), 13.27 (E) (d, $J=1.0\text{Hz}$, CH_3CH), 15.77 (d, $J=6.5\text{Hz}$, $2\times\text{CH}_3\text{CH}_2$), 65.12 (d, $J=6.5\text{Hz}$, $2\times\text{CH}_2$), 112.21 (E), 113.84 (Z) (2d, $J=1.5, 2.5\text{Hz}$, respectively, COP), 121.27 (Z), 122.00 (E) (2d, $J=9.0$ and 10.0, respectively, CN), 130.96 (E), and 131.52 (Z) (2d, $J=5.5$ and 6.9, respectively, CH); δ_{P} -5.5 (E-isomer) and -5.7 (Z-isomer); m/z 219 (M^+ , 4%), 163 (64), 109 (40), 108 (20), 99 (43), 91 (18), 82 (35), 81 (100), 69 (10), 66 (21), 65 (20), 56 (61), 55 (41), 54 (23), 47 (10), 45 (11), and 43 (10). Anal. Calcd. for $\text{C}_8\text{H}_{14}\text{NO}_4\text{P}$: C, 43.84; H, 6.44; N, 6.39. Found: C, 44.03; H, 6.74; N, 6.66.

2-(Diethylphosphonoxy)-3-pentenitrile (2b): R_f 0.49; ν_{\max} 1270 (P=O) and 1100-940 cm^{-1} (P-O)^{4b}; δ_{H} 1.35, 1.38 (2t, $J=7.0\text{Hz}$, 6H, $2\times\text{CH}_3\text{CH}_2$), 1.82 (d, $J=6.5\text{Hz}$, 3H, CH_3CH), 4.16 (m, 4H, $2\times\text{CH}_2\text{O}$), 5.46 (ddd, $J=8.0, 7.0, 1.0\text{Hz}$, 1H, CHO), 5.64 (ddq, $J=15.0, 7.0, 2.0\text{Hz}$, 1H, CHCHO), and 6.17 (ddq, $J=15, 6.5, 2.0\text{Hz}$, 1H, CHCH_3)^{4b}; δ_{C} 15.71 (d, $J=7.0\text{Hz}$, $2\times\text{CH}_3$), 17.28 (CH_3CH), 64.35, 64.42 (2d, $J=6.0\text{Hz}$, $2\times\text{CH}_2\text{O}$), 64.72 (d, $J=5.0\text{Hz}$, CHCN), 115.47 (d, $J=6.0\text{Hz}$, CN), 122.23 (CHCHO), and 135.24 (CHCH_3); m/z 234 (M^++1 , 0.2%), 233 (M^+ , 1.2), 177 (29), 127 (20), 125 (11), 99 (100), 81 (29), 80 (15), 79 (11), and 53 (15).

2-(Diethylphosphonoxy)-3,3-dimethylbutanenitrile (2c): R_f 0.56; ν_{\max} 1270 (P=O), 1020, and 980 cm^{-1} (P-O); δ_{H} 1.12 (s, 9H, $3\times\text{CH}_3\text{C}$), 1.38, 1.39 (2t, $J=7.0\text{Hz}$, $2\times\text{CH}_3\text{CH}_2$), 4.20 (m, 4H, $2\times\text{CH}_2$), and 4.65 (d, $J=8.5\text{Hz}$, CH); δ_{C} 15.73, 15.76 (2d, $J=6.5\text{Hz}$, $2\times\text{CH}_3\text{CH}_2$), 24.59 ($3\times\text{CH}_3\text{C}$), 35.43 (d, $J=6.0\text{Hz}$, CCH), 64.43, 64.44 (2d, $J=6.0\text{Hz}$, $2\times\text{CH}_2$), 73.19 (d, $J=7.0\text{Hz}$, CH), and 115.82 (d, $J=2.0\text{Hz}$, CN); δ_{P} -1.4; m/z 249 (M^+ , 0.1%), 209 (13), 193 (46), 178 (15), 165 (28), 155 (32), 137 (100), 127 (62), 125 (22), 99 (78), 82 (24), 81 (43), 57 (23), 53 (12), and 41 (24).

2-(Diethylphosphonoxy)-2-phenylethanenitrile (2d): R_f 0.49; ν_{\max} 1265 (P=O), 1030, and 970 cm^{-1} (P-O)^{6a}; δ_{H} 1.22, 1.38 (2t, $J=7.0\text{Hz}$, 6H, $2\times\text{CH}_3$), 4.01, 4.17 (2m, 4H, $2\times\text{CH}_2\text{O}$), 6.08 (d, $J=9.0\text{Hz}$, CHO), 7.45, and 7.57 (2m, 5H, Ph)^{6a}; δ_{C} 15.76, 15.72 (2d, $J=7.0\text{Hz}$, $2\times\text{CH}_3$), 64.41, 64.59 (2d, $J=6.0\text{Hz}$, $2\times\text{CH}_2$), 66.25 (d, $J=4.5\text{Hz}$, CHO), 115.93 (d, $J=6.0\text{Hz}$, CN), 127.25, 129.00, 130.31 ($3\times\text{CH-Ph}$), and 132.27 (d, $J=$

5.5Hz, C₁-Ph); *m/z* 270 (*M*⁺+1, 0.2%), 269 (*M*⁺, 17), 213 (57), 125 (10), 117 (10), 116 (100), 115 (60), 105 (45), 99 (12), 89 (44), 81 (23), 77 (19), and 63 (14).

2-(Diethylphosphonoxy)-3-phenylpropanenitrile (2e): *R_f* 0.44; *v*_{max} 1270 (P=O), 1030, and 970 cm⁻¹ (P-O); δ_H 1.24-1.33 (m, 6H, 2xCH₃), 3.22 (d, *J* = 7.0Hz, 2H, CH₂C), 3.94-4.16 (2m, 4H, 2xCH₂O), 5.16 (dd, *J* = 15.0 y 7.0Hz, 1H, CHO), 7.32 (m, 5H, Ph); δ_C 15.70 (d, *J* = 6.5Hz, 2xCH₃), 40.15 (d, *J* = 6.0Hz, CH₂C), 64.40, 64.48 (2d, *J* = 6.0Hz, 2xCH₂O), 65.19 (d, *J* = 6.0Hz, CHO), 116.27 (d, *J* = 4.0Hz, CN), 127.70, 128.58, 129.46 (3xCH-Ph), and 132.80 (C₁-Ph); *m/z* 283 (*M*⁺, 0.2%), 130 (15), 129 (100), and 91 (35).

2-(Diethylphosphonoxy)-2-methyl-3-butenenitrile (2g): *R_f* 0.44; *v*_{max} 3080, 1660 (CH=C), 2240 (CN), 1265 (P=O), and 1080-950 cm⁻¹ (P-O)^{4b}; δ_H 1.36, 1.37 (2t, *J* = 7.0Hz, 6H, 2xCH₃CH₂), 1.90 (s, 3H, CH₃CO), 4.17 (m, 4H, 2xCH₂O), 5.47 (d, *J* = 10.5Hz, 1H, CH₂=C), 5.74 (d, *J* = 17.0Hz, 1H, CH₂=C), and 6.06 (dd, *J* = 17.0, 10.5Hz, 1H, CH)^{4b}; δ_C 15.70, 17.79 (2d, *J* = 7.0Hz, 2xCH₃CH₂), 27.73 (d, *J* = 5.0Hz, CH₃CO), 64.24, 64.32 (2d, *J* = 6.0Hz, 2xCH₂O), 74.15 (d, *J* = 7.0Hz, CO), 117.01 (d, *J* = 4.0Hz, CN), 118.62 (CH₂=C), and 134.57 (d, *J* = 4.5Hz, CH); *m/z* 233 (*M*⁺, 0.8%), 177 (21), 127 (42), 125 (19), 109 (20), 99 (100), 97 (40), 82 (13), 81 (41), 80 (28), 79 (16), and 53 (20).

2-Ethyl-2-(diethylphosphonoxy)butanenitrile (2h): *R_f* 0.44; *v*_{max} 1270 (P=O) and 1020 cm⁻¹ (P-O); δ_H 1.12 (t, *J* = 7.5Hz, 6H, 2xCH₃CH₂C), 1.37 (t, *J* = 7.0Hz, 2xCH₃CH₂O), 2.08 (m, 4H, 2xCH₂C), and 4.18 (m, 4H, 2xCH₂O); δ_C 8.04 (2xCH₃CH₂C), 15.82 (d, *J* = 6.5Hz, 2xCH₃CH₂O), 31.29 (d, *J* = 4.0Hz, 2xCH₂C), 64.22 (d, *J* = 6.0Hz, 2xCH₂O), 79.80 (d, *J* = 8.0Hz, CO), and 117.76 (d, *J* = 5.5Hz, CN); δ_P -5.4; *m/z* 234 (*M*⁺-15, 0.7%), 164 (15), 155 (41), 127 (63), 109 (13), 99 (100), 81 (26), and 57 (13). Anal. Calcd. for C₁₀H₂₀NO₄P: C, 48.19; H, 8.09; N, 5.60. Found: C, 48.60; H, 7.80; N, 5.75.

cis/trans-1-(Diethylphosphonoxy)-1,3-cyclopentanedicarbonitrile (4i): *v*_{max} 2240 (CN), 1270 (P=O), and 1000 cm⁻¹ (P-O); δ_H 1.38 (t, *J* = 7.0Hz, 6H, 2xCH₃), 2.63 (m, 6H, 3xCH₂C), 3.28 (m, 1H, CH), and 4.21 (m, 4H, 2xCH₂O); δ_C 15.77 (d, *J* = 6.5Hz, 2xCH₃), 25.27, 25.39 (CH), 27.74, 28.20 (CH₂CH₂CH), 38.83, 39.15 (2d, *J* = 5.0 and 4.5Hz, respectively, OCCH₂CH₂), 43.48, 43.55 (2d, *J* = 5.5 and 5.0Hz, respectively, OCCH₂CH), 64.60, 64.76 (2d, *J* = 5.5 and 6.0, respectively, CH₂O), 104.25, 104.32 (CO), 116.85 (d, *J* = 1.5Hz, OCCN), 117.14 (OCCN), 120.42 and 120.43 (CHCN); δ_P -4.7 and -4.9; *m/z* 271 (*M*⁺-1, 0.6%), 127 (39), 125 (12), 99 (100), 92 (15), 91 (82), 82 (15), 81 (45), 65 (19), 64 (15), and 63 (12). Anal. Calcd. for C₁₁H₁₇N₂O₄P: C, 48.53; H, 6.29; N, 10.29. Found: C, 48.09; H, 6.57; N, 10.41.

*1-(Diethylphosphonoxy)cyclohexanecarbonitrile (2j)*⁸: *R_f* 0.47; *v*_{max} 1270 (P=O), 1030, and 1000 cm⁻¹ (P-O); δ_H 1.37, 1.38 (2t, *J* = 7.0Hz, 6H, 2xCH₃), 1.64, 1.84, 1.97, 2.30 (4m, 10H, 5xCH₂C), and 4.18 (m, 4H, 2xCH₂O); δ_C 15.79 (d, *J* = 7.0Hz, 2xCH₃), 22.30, 23.89 (3xCH₂), 36.81 (d, *J* = 5.0Hz, 2xCH₂CO), 64.18 (d, *J* = 6.0Hz, 2xCH₂O), 75.59 (d, *J* = 8.0 Hz, CO), and 118.34 (d, *J* = 3.5Hz, CN); *m/z* 246 (*M*⁺-15, 0.2%), 155 (43), 127 (58), 94 (100), 81 (31), and 41 (17).

1-(Diethylphosphonoxy)-4-oxocyclohexanecarbonitrile Ethylene Ketal (2k): *R_f* 0.31; *v*_{max} 1270 (P=O) and 1000 cm⁻¹ (P-O); δ_H 1.38 (t, *J* = 7.0Hz, 2xCH₃), 1.83 (m, 4H, 2xCH₂CO), 2.33 (m, 4H, 2xCH₂CCN), 3.96 (s, 4H, OCH₂CH₂O), and 4.18 (m, 4H, CH₂CH₃); δ_C 15.87 (d, *J* = 7.0Hz, 2xCH₃), 30.37 (2xCH₂CO), 34.38 (d, *J* = 5.0Hz, 2xCH₂CCN), 64.38, 64.42 (OCH₂CH₂O), 64.41 (d, *J* = 5.5Hz, 2xCH₂OP), 73.48 (d, *J* = 8.0Hz, COP), 106.16 (COCH₂), and 118.18 (d, *J* = 2.5Hz, CN); δ_P -4.9; *m/z* 291 (*M*⁺-28, 1.1%), 163 (38),

99 (65), 86 (100), and 81 (11).

2-(Diethylphosphonoxy)-2-phenylpropanenitrile (2l)2a: R_f 0.41; ν_{\max} 2220 (CN), 1270 (P=O), 1030, and 980 cm^{-1} (P-O); δ_{H} 1.28, 1.32 (2t, $J = 7.0\text{Hz}$, 6H, $2 \times \text{CH}_3\text{CH}_2$), 2.14 (s, 3H, CH_3CO), 4.09 (m, 4H, $2 \times \text{CH}_2$), 7.46, and 7.64 (2m, 5H, Ph); δ_{C} 15.70, 15.72 (2d, $J = 7.0\text{Hz}$, $2 \times \text{CH}_3\text{CH}_2$), 30.13 (d, $J = 5.0\text{Hz}$, CH_3CO), 64.22, 64.31 (2d, $J = 6.5\text{Hz}$, $2 \times \text{CH}_2$), 75.44 (d, $J = 8.0\text{Hz}$, CO), 118.15 (CN), 124.91, 128.65, 129.48 ($3 \times \text{CH-Ph}$), and 137.73 (d, $J = 5.5\text{Hz}$, C-Ph); m/z 284 (M^{+1} , 1%), 283 (M^+ , 13), 155 (15), 147 (70), 131 (12), 130 (56), 129 (100), 128 (15), 127 (32), 125 (14), 105 (19), 103 (52), 102 (25), 99 (58), 91 (10), 82 (10), 81 (35), 78 (14), 77 (48), and 51 (15).

2-(Diethylphosphonoxy)bicyclo[2.2.1]heptane-2-carbonitrile (2m): R_f 0.58; ν_{\max} 2225 (CN), 1280 (P=O), and 1025 cm^{-1} (P-O); δ_{H} 1.38 (m, 6H, $2 \times \text{CH}_3$), 1.57, 1.87, 2.36 (3m, 7H, CH_2CH_2 and CHCH_2CO), 2.39 (d, $J = 3.0\text{Hz}$, 2H, CH_2CO), 2.89 (m, CHCO), and 4.20 (m, 4H, $2 \times \text{CH}_2\text{O}$); δ_{C} 15.79 (d, $J = 7.0\text{Hz}$, $2 \times \text{CH}_3$), 20.75, 27.93, 37.20 (CH_2CH_2 and CHCH_2CH), 39.09 (CHCH_2CO), 44.36 (d, $J = 4.5\text{Hz}$, CH_2CO), 48.15 (d, $J = 5.0\text{Hz}$, CHCO), 64.20, 64.27 (2d, $J = 5.5\text{Hz}$, $2 \times \text{CH}_2\text{O}$), 77.99 (d, $J = 9.0\text{Hz}$, COP), and 119.95 (CN); δ_{P} -4.0; m/z 273 (M^+ , 0.3%), 206 (14), 155 (100), 150 (21), 127 (37), 99 (54), 91 (19), 81 (32), 67 (16), 65 (12), and 41 (13). Anal. Calcd. for $\text{C}_{12}\text{H}_{20}\text{NO}_4\text{P}$: C, 52.74; H, 7.38; N, 5.13. Found: C, 52.32; H, 7.29; N, 5.12.

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